





Figure 1: Water Bath Configuration for Performing the J&L Modified Kawasaki Condensate Corrosion test.

Once the bath was configured, the samples were suspended without touching the bottom or the sides of the beakers. The beakers were filled with the corrosive solution that was used in the Kawasaki Condensate Corrosion Test [3]. The chemical composition is shown in Table 1.

Table 1 Chemical Composition of the Kawasaki Condensate Corrosion Test Solution (ppm)

Cl <sup>-</sup>	SO <sub>3</sub> <sup>2-</sup>	SO <sub>4</sub> <sup>2-</sup>	CO <sub>3</sub> <sup>2-</sup>	NO <sub>2</sub> <sup>-</sup>	NO <sub>3</sub> <sup>-</sup>	CH <sub>3</sub> COO <sup>-</sup>	HCHO	COOH <sup>-</sup>	NH <sub>4</sub> <sup>+</sup>	Activated carbon
50	250	1250	2000	100	20	400	250	100	2500	50g/l

Beakers remained in the water bath until the corrosive solution evaporated (approximately 26 hours). Samples were then rinsed and cleaned by scrubbing gently with a paintbrush. This cycle was repeated 10 times. After the final cycle, the samples were once again cleaned. The samples were photographed to show the corrosion products that had built up from the test. These samples were cleaned using an 8% HNO<sub>3</sub> solution. Three sets of samples were tested for repeatability.

## Results:

Table 2: J&L Modified Kawasaki Condensate Corrosion Test Results: Maximum Pit Depth in Substrate (µm).

Set Number	Bare 409	Aluminized 409	Neukote™ 409	Bare 439	Neukote™ 439
Set 1	140	159	80 <sup>2</sup>	145	20 <sup>3</sup>
Set 2	209	>630 <sup>1</sup>	57 <sup>2</sup>	151	40 <sup>3</sup>
Set 3	179	>630 <sup>1</sup>	48 <sup>2</sup>	157	79 <sup>3</sup>
Maximum	209	>630 <sup>1</sup>	80 <sup>2</sup>	157	79 <sup>3</sup>

<sup>1</sup> 409 Thickness using an average aluminized coating thickness of 15 µm.

<sup>2</sup> 409 Thickness using an average Neukote™ coating thickness of 18 µm.

<sup>3</sup> 439 Thickness using an average Neukote™ coating thickness of 20 µm.

## Discussion:

The samples will be discussed in order of increasing pitting corrosion resistance. The Aluminized 409 (Al409) had poor pitting resistance. In all three sets, the Al409 did the worst of all the samples tested. In the first set it had pitting up to 159 micrometers. In the second and third sets the samples perforated (aluminized samples were 630 micrometers thick) (see Figure 2).

The uncoated 409 corroded the most in all three runs except the first one (aside from the aluminized 409) in which the uncoated 439 corroded 5 micrometers more. The corrosion shown in both the bare 409 and the

bare 439 were very similar in looks with the biggest difference being that there was more corrosion on the 409 than the 439 (see Figure 3).



Figure 2: Al409 with corrosion removed.

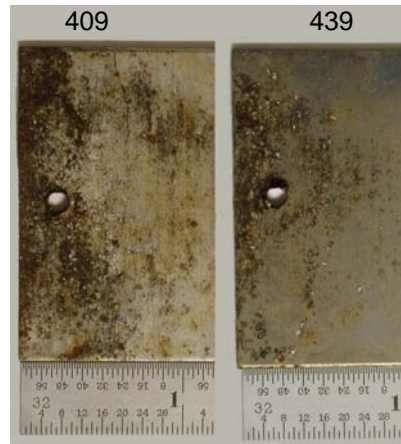


Figure 3: Uncoated 409 and 439 after test

The last two samples were the Neukote™ samples. These samples had less pitting than the bare and Al409 samples in all three sets. Neukote™ 439 performed approximately the same as Neukote™ 409. Several things were noticed during the corrosion removal process for these coatings. The Neukote™ coating was extremely difficult to remove. The nitric solution and ultrasonic cleaning had a minimal effect on the removal of the coating and the corrosion. What was discovered, was that the Neukote™ formed small pimples all over the sample (see Figure 4).



Figure 4: Neukote™ sample - pimples after test completion on the left and pimples removed on the right (approx. 7x).

This is thought to have happened due to the sulfur that was present in the test solution [4]. Pitting occurred underneath these pimples. It should be noted, corrosion was only found under a small percentage of these pimples. When looking at the samples from a distance, these pimples were not noticeable.

#### Conclusion:

Neukote™, provided better pitting resistances than any of the other samples that were tested. The maximum pit depth on the Neukote™ samples were 80 and 79 micrometers for 409 and 439 respectively. Compared to uncoated 409 and 439 that had pitting depths of 209 and 157 micrometers respectively. The other coatings that were tested also performed worse than the Neukote™. The Aluminized 409 became perforated in sets two and three. When it came to appearance the Neukote™ also out performed the other samples. When standing back 15 feet from the samples, the Neukote™ samples exhibited minimal corrosion. All of the other samples had noticeable corrosion even from this distance. The J&L Modified Kawasaki Condensate Corrosion test has shown that Neukote™ is the best solution for pitting resistance for exhaust systems.

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References:

1. R. R. Gaugh, "Corrosion of Materials for Automotive Exhaust Systems," SAE Technical Paper, p.305, 912294 (1991)
2. T. Ujiro, M. Kitazawa and F. Togashi, "Corrosion of Muffler Materials in Automotive Exhaust Gas Condensates", Materials Performance, Vol.33, No.12, p.49 (1994)
3. J. Hirasawa, T. Ujiro, S. Satoh and O. Furukimi, "Development of High Corrosion Resistant Stainless Steels for Automotive Mufflers Based on Condensate Corrosion Test and Field Investigation", SAE Technical Paper, SP-1614 (2001).
4. Conversation with Dr. James E. Neely, inventor of Neukote™ on February 19, 2003.